owned by the bank holding company, including the companies listed below.

The applications listed below, as well as other related filings required by the Board, are available for immediate inspection at the Federal Reserve Bank indicated. The applications will also be available for inspection at the offices of the Board of Governors. Interested persons may express their views in writing on the standards enumerated in the BHCA Act (12 U.S.C. 1842(c)). If the notice or the offices of the Board of Governors. Interested persons may express their views in writing to the Reserve Bank indicated for that notice or to the offices of the Board of Governors. Comments must be received not later than March 20, 2017.

Federal Reserve Bank of Minneapolis
(Jacquelyn K. Brunmeier, Assistant Vice President) 90 Hennepin Avenue, Minneapolis, Minnesota 55480—0291:
1. Alerus Financial Employee Stock Ownership Plan, Grand Forks, North Dakota; to acquire additional shares of Alerus Financial Corporation, Grand Forks, North Dakota, and indirectly acquire additional shares of Alerus Financial, National Association, Grand Forks, North Dakota.

Yao-Chin Chao,
Assistant Secretary of the Board.

DEPARTMENT OF HEALTH AND HUMAN SERVICES
Centers for Disease Control and Prevention
[Docket Number CDC–2017–0015, NIOSH–295]
Health Risks to Workers Associated With Occupational Exposures to Peracetic Acid; Request for Information
AGENCY: National Institute for Occupational Safety and Health (NIOSH) of the Centers for Disease Control and Prevention (CDC), Department of Health and Human Services (HHS).
ACTION: Request for information.
SUMMARY: The National Institute for Occupational Safety and Health of the Centers for Disease Control and Prevention intends to evaluate the scientific and technical data on occupational exposures to peracetic acid (CAS #79–21–0, also known as peroxyacetic acid and PAA). NIOSH is requesting information on the following: (1) Workplace exposure data for peracetic acid, (2) possible health effects observed in workers exposed to peracetic acid, (3) workplaces and products in which peracetic acid may be found, (4) description of work tasks and scenarios with a potential for exposure to peracetic acid, (5) reports and findings from in vitro and in vivo toxicity studies with peracetic acid, (6) data applicable to the quantitative risk assessment of health effects associated with acute, subchronic and chronic workplace exposures to peracetic acid, (7) sampling and analytical methods for peracetic acid, and (8) control measures, including engineering controls, work practices, and personal protective equipment (PPE), that are being used in workplaces where there is potential for exposure to peracetic acid.
DATES: Electronic or written comments must be received by June 5, 2017.
ADDRESSES: You may submit comments, identified by CDC–2017–0015 and docket number NIOSH–295, by any of the following methods:
• Federal eRulemaking Portal: http://www.regulations.gov. Follow the instructions for submitting comments.
• Mail: National Institute for Occupational Safety and Health, NIOSH Docket Office, 1090 Tusculum Avenue, MS C–34, Cincinnati, Ohio 45226–1996.
Instructions: All information received in response to this notice must include the agency name and docket number [CDC–2017–0015; NIOSH–295]. All relevant comments received will be posted without change to www.regulations.gov, including any personal information provided. For access to the docket to read background documents or comments received, go to www.regulations.gov.
FOR FURTHER INFORMATION CONTACT: G. Scott Dotson, NIOSH, Education and Information Division, Robert A. Taft Laboratories, 1090 Tusculum Avenue, Cincinnati, OH 45226, (513) 533–8540 (not a toll free number).
SUPPLEMENTARY INFORMATION: Peracetic acid is a peroxide-based molecule used extensively as an antimicrobial agent in many commercial applications. It is routinely used as a sterilant during the cleaning of endoscopes and other medical devices, as a disinfectant in food processing, as a bleaching agent, and in the synthesis of other chemicals [NAS 2010; Pechacek et al. 2015]. The chemical and physical properties of peracetic acid make the molecule highly reactive, unstable, and volatile. Peracetic acid has a pungent, vinegar-like odor [NAS 2010]. Peracetic acid is formed from a sulfuric acid-catalyzed chemical reaction between acetic acid and hydrogen peroxide [NAS 2010]. Peracetic acid solutions typically consist of a mixture of peracetic acid,